

Biochemistry
Course Code: BME 291
Level: 200
Allowed Time: 2 hours
Second Semester 2015 /2016

nd Semester 2015 /2016 Final term exam 50 marks



### Biochemistry exam

#### تعليمات

- الإجابة في الأماكن المخصصة فقط في نفس ورقة الامتحان وكل سؤال على حدة.
  - ممنوع تكرار أية إجابة أو إعادتها في أماكن أخرى.
    - ليست هناك أية فرصة لإضافة ورق زائد.
  - الصفحة الأخيرة مسودة ولا يعتد بأي كتابة فيها كإجابة.
- يتكون الإمتحان من خمسة اسئلة ويقع في 8 صفحات بالمسودة وجميع الأسئلة إجبارية.
- ممنوع أية كتابات خارجة عن مضمون الإجابة حتى لا تعرض نفسك للمسائلة القانونية.

#### Directions

- All questions are to be attempted in the same exam papers.
- Answers should be written in the provided spaces.
- Do not repeat any answer in other places.
- No additional Booklets could be provided.
- The last paper is a Draft Paper not to be corrected.
- The exam consists of 5 questions in 8 pages including a draft page.



Biomedical Engineering

Biochemistry Course Code: BME 291 Level: 200

Allowed Time: 2 hours Second Semester 2015 /2016 Final term exam

50 marks



Faculty Of Engineer 04-06-2016

# Question (1): Choose the correct answer and write the letter of your choice in the answer table below:

(7 marks, 1 mark/eacl

1- Oxidation of the first hydroxy group of glucose results a-gluconic acid	in the formation of:
c- glucaric acid	b- glucuronic acid d- galactonic acid
2- The branched part of starch is known as:	

a- maltose

b- cellibiose c- amylose d- amylopectin

3- Regarding the biological value, lysine is classified as:

a- glucogenic amino acid b- ketogenic amino acid c- both glucogenic and ketogenic amino acid d- essential amino acid

4- The c terminal of "valyl leucyl glycyl arginine" peptide is:

a- valine b- glycine c- arginine d- leucine

5- An example of nucleoside in DNA structure is:

a- adenine and ribose b- cytosine and ribose c- uracil and deoxyribose d- thymine and deoxyribose

6- Lignoceric acid is an example of fatty acid which contains:

a-1 double bond b- 2 double bonds c- 3 double bonds d- no double bonds

7- An example of 6C fatty acid is:

a- butyric acid b- palmitic acid c- caproic acid d- stearic acid

Question	1	2	3	4	5	6	7
Answer letter					-	-	-



each)

Biochemistry Course Code: BME 291 Level: 200

Allowed Time: 2 hours Second Semester 2015 /2016 Final term exam

Faculty Of Engineering 04-06-2016

50 marks

Question (2): Choose the correct answer for statements in column A from column B and write the answer letter in answer column: (14 marks, 1 mark/each)

No	Column A	Answer		Column B
1	Branched chain amino acid		A	γ-Glutamyl-
2	Amino acid with hydroxy group			cysteinyltyrosine
3	Amino acid with hydroxy group		В	lock and key model
4			C	valine
	The process of releasing neurotransmitters from vesicles		D	zymogen
5	The enzyme molecule changes its shape as the substrate molecules gets close	102-20012 10	E	tyrosine
6	Conjugated protein enzymes		F	phenylalanine
7	Materials used in the production of lactose free milk	new man parti	G	holoenzyme
8	Glutathione		Н	cardiolipin
9	Production of unpleasant odour and taste from fats.		I	LDL
10	Good cholesterol		J	exocytosis
11	The protein component of lipoproteins		K	apoprotein
12	Phospholipid containing 3 molecules of glycerol		L	guanine
13	The nitrogen base that bound to thymine in DNA double helix.		M	γ-Glutamyl- cysteinylglycine
14	connects all parts of the body and provides support		N	arginine
	Company of the National Company		0	HDL
			P	rancidity
			Q	connective tissue
			R	plasmalogen
			S	immobilized enzymes
			T	adenine
			U	induced-fit model

14



Biomedical Engineering

Biochemistry Course Code : BME 291 Level : 200

Allowed Time : 2 hours Second Semester 2015 /2016

Final term exam 50 marks



Faculty Of Eng 04-06-2016

## Question (3): Give the name for each of the followings:

(16 marks, 1 mark/

No.	Statement	(10 marks, 1 mar
1		Answer
	The polysaccharide that contains $\beta$ glycosidic linkage between $\beta$ glucose units.	
2	The optically inactive amino acid.	
3	The disaccharide that is composed of galactose and glucose units.	
4	Sugar isomers that are mirror images of each other.	
5	The gap between different nerve cells.	
6		
	Protein that is composed mainly of hydroxyproline and hydroxylysine.	
7	Receptors that are responsible for slow-acting neurotransmitters.	
8	The breakdown of the 1ry protein structure.	
9	The number of substrate molecules converted into products per second per	
	enzyme molecule.	
0	Organic molecules that are required by certain enzymes to carry out catalysis.	
1	A molecule is similar to the substrate and competes with the substrate for the	
	active site.	
2	The product of lipid hydrolysis that has detergent action.	
	The process of margarine production	
	The RNA type that carries the amino acid to be incorporated into the developing protein.	
5	The substrate concentration at which the reaction velocity is equal to ½ Vmax.	
1	A group of two or more different types of tissues that work together to perform a specific function.	



ering

Biochemistry Course Code : BME 291 Level: 200

Allowed Time: 2 hours Second Semester 2015 /2016 Final term exam



Faculty Of Engineering 04-06-2016

	50 marks	21.002,010
uestion (4): Determine whether the follo	owing sentences are correct or wrong and	d correct underlined
ords in case of wrong statements:	The state of the s	
		(7 marks, 1 mark/eac
1- The carboxy group of amino acid rem  Answer:	ains unionized in <u>alkaline medium</u> .	nighten sammen series
2- All DNA strands are read from the 5' Answer:	to the 3' end.	
3- <u>Deploarization</u> has inhibitory effect o <u>Answer:</u>	n nerve impulse.	
o me vitalli.	vay acts as a <u>noncompetitive Inhibitor</u> to	one of the enzymes
Answer:  5- The characteristic secondary structure of Answer:	of collagen is <u>β-pleated sheets</u> .	
Manual Co.	eases rate of enzyme reaction due to satura	tion of enzyme
7- The individual phospholipids of cell men	mbrane has <u>two dimensional movement</u> .	



Biomedical Engineering

Biochemistry Course Code : BME 291 Level : 200

Allowed Time : 2 hours Second Semester 2015 /2016

Final term exam 50 marks



Faculty Of Engineering 04-06-2016

	(6 marks)
A- Complete the following sentences:	(4 marks, 0,5 mark/each)
1- Membrane phospholipid bilayer is composed of	a head that is regarding its
solubility in water.	
2- The buffer that maintains membrane fluidity is	
3- At higher temperatures,	
	usion across cell membrane such as
5- Solution with concentration greater than that ins	ide the cell is known as
6- The two types of transport across cell membrane	that require transport proteins are
and	1
B- Give the name for all the following compound	ds: (2 marks, 0,5 mark/each)
1-	
	( , o,c mario cacii)



Biochemistry

Course Code : BME 291

Level: 200 Allowed Time: 2 hours

4-

Second Semester 2015 /2016 Final term exam 50 marks



Faculty Of Engineering 04-06-2016

CH3
S
CH2
CH2
CH2
H2N—C—COOH

Name of the compound: .....

H<sub>2</sub>N-CH-C-OH CH<sub>2</sub> CH<sub>2</sub> CH<sub>2</sub> C=O

Name of the compound:

6

**End of Questions** 

Examination Committee

Dr. Mohamed El-Mesery