

Mansoura University
Faculty of Engineering
BME & MTE Programs
Course Title: Thermodynamics
Course Code: MPE 172



Level: 100
Exam Type: Final
Date: 17 /12/2017
Time: 2 Hours
Full Mark: 50

Answer all the following questions. Use of steam tables are allowed

Question (1) [12 Marks]

0.5 m³ of air at 300 K and 1 bar is heated at constant volume to 5 bar. It is then cooled at constant pressure and further expanded isothermally so as to reach the condition from where it started. Sketch the cycle on P-v diagram and determine :

- (i) Temperature at the end of constant volume heating.*
- (ii) Heat, work and change in entropy for each process.*
- (iii) Net work done and heat transferred during the cycle.*

Take $C_p = 1005 \text{ J/kg.K}$ and $C_v = 718 \text{ J/kg.K}$.

Question (2) [10 Marks]

Steam flows steadily through an adiabatic nozzle. The inlet conditions of the steam are 10 bar , 250°C, and 80 m/s, and the exit conditions are 1.5 bar and dry saturated steam. The mass flow rate of the steam is 3 kg/s. Determine (i) the velocity of the steam leaving the nozzle , (ii) the exit area of the nozzle and (iii) the rate change in entropy.

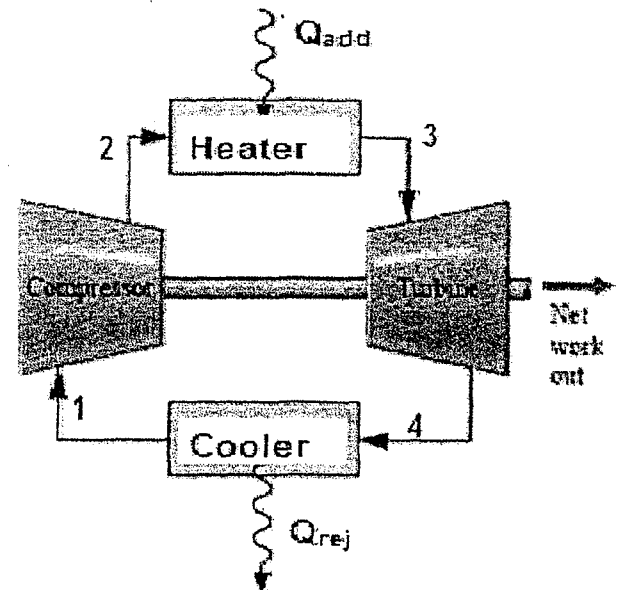
Question (3) [10 Marks]

Consider a building whose annual air-conditioning load is estimated to be 120,000 kWh in an area where the unit cost of electricity is \$0.10/kWh. Two air conditioners are considered for the building. Air conditioner "A" has a seasonal average COP of 3.2 and costs \$5500 to purchase and install. Air conditioner "B" has a seasonal average COP of 5.0 and costs \$7000 to purchase and install. All else being equal, determine which air conditioner is a better buy. Assume the average life time of air conditioner is 10 years.

Question (4) [10 Marks]

In a gas turbine plant working on Brayton cycle, the air at inlet is 27 °C, 1 bar . The pressure ratio is 9 and the maximum temperature is 900 °C. Sketch the cycle on T-s diagram and determine:

- The thermal efficiency of the cycle.
- The back work ratio.
- The required mass flow rate if the net output power is 100 MW.



Question (5) [8 Marks]

Moist air at the standard atmospheric pressure, has a temperature of 30 oC and a relative humidity 50% . Determine the following:

- the partial pressure of dry air and the water vapor,
- the specific humidity,
- the degree of saturation,
- the dew point temperature; and
- the density of moist air.

Notes

Process	Q	W	P, v, T relation
Constant volume	$m C_v(T_2 - T_1)$	0	$\frac{T_2}{T_1} = \frac{P_2}{P_1}$
Isothermal	$P_1 V_1 \ln(V_2/V_1)$	$P_1 V_1 \ln(V_2/V_1)$	$\frac{P_2}{P_1} = \frac{v_2}{v_1}$
Polytropic	$(\gamma - n) W / (\gamma - 1)$	$(P_1 V_1 - P_2 V_2) / (n - 1)$	$\left(\frac{T_2}{T_1}\right)^{\frac{\gamma}{\gamma - 1}} = \left(\frac{P_2}{P_1}\right)^{\frac{\gamma}{\gamma - 1}} = \left(\frac{v_1}{v_2}\right)^{\frac{\gamma}{\gamma - 1}}$

Entropy change during thermodynamic processes for ideal gas

$$\Delta S = m \left[c_v \ln \left(\frac{T_2}{T_1} \right) + R \ln \left(\frac{v_2}{v_1} \right) \right] = m \left[c_p \ln \left(\frac{T_2}{T_1} \right) - R \ln \left(\frac{p_2}{p_1} \right) \right] = m \left[c_p \ln \left(\frac{v_2}{v_1} \right) + c_v \ln \left(\frac{p_2}{p_1} \right) \right]$$

The enthalpy of the moist air is given by:

$$h = 1.005 T + \omega (2500.9 + 1.88 T) \left(\frac{kJ}{kg} \text{ dry air} \right)$$

Good Luck
Dr. Ahmed Abd-El salam